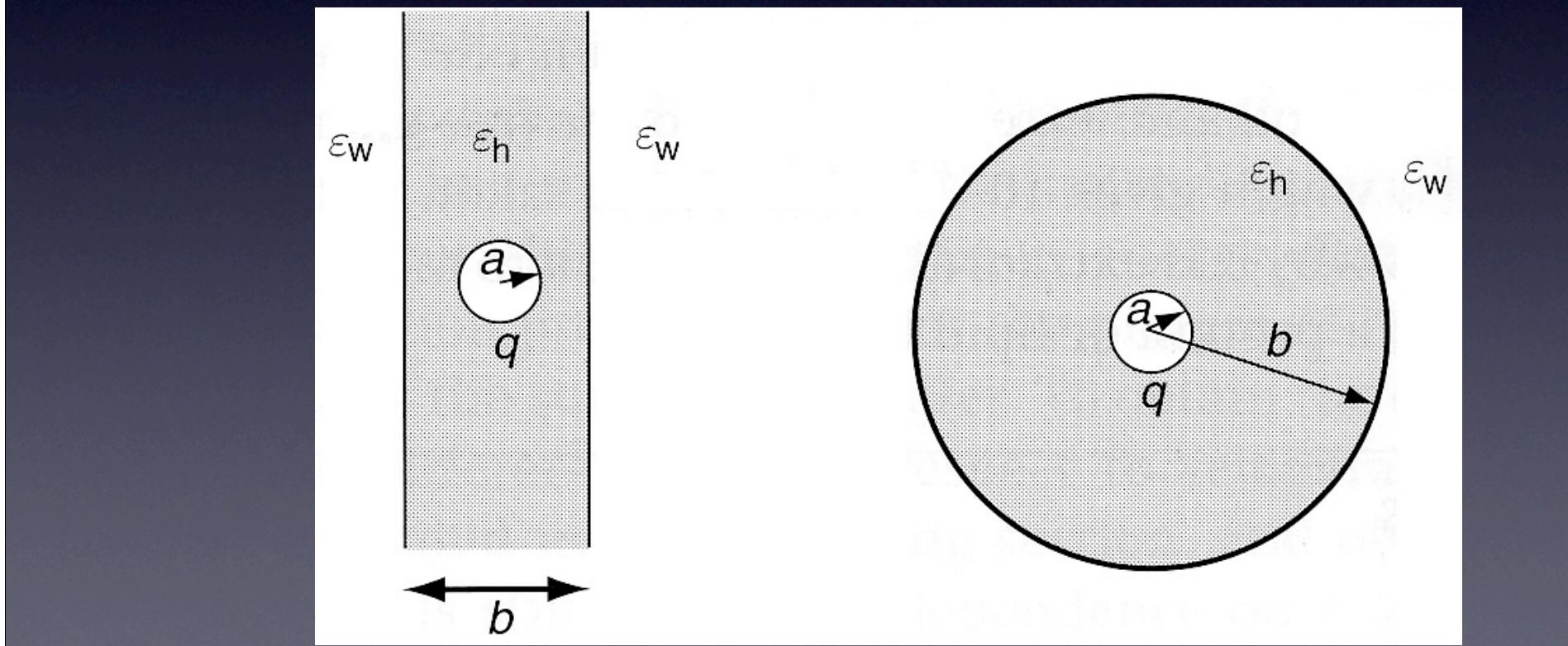
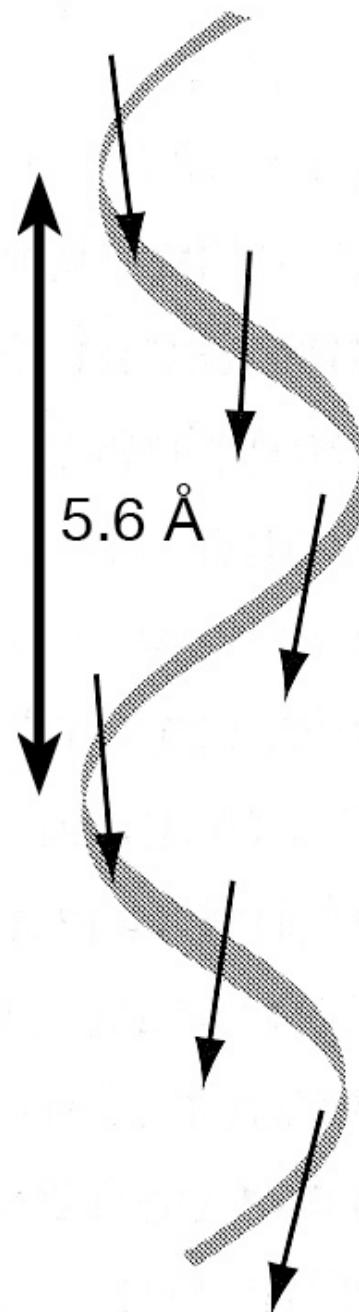
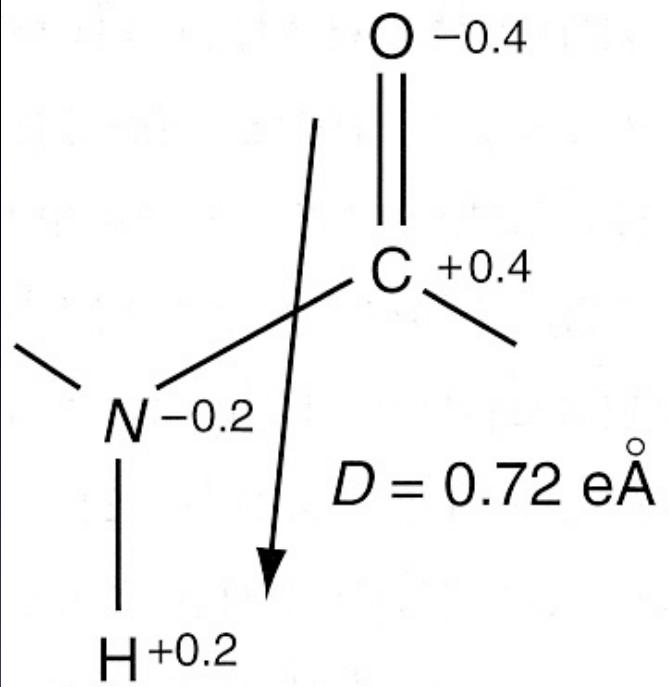


image forces

dielectric self energy





dipolar moment
of an α -helix

	polarizability $\alpha/4\pi\epsilon_0$ (10^{-30} m^3)	perm. dipole moment μ (D)	Keesom $U_{\text{Keesom}} \times r^6$ (10^{-79} Jm^6)	Debye $U_{\text{Debye}} \times r^6$ (10^{-79} Jm^6)	London $U_{\text{London}} \times r^6$ (10^{-79} Jm^6)	Tot. vdW theoretical (10^{-79} Jm^6)	energy as obtained from fit to gas eqn. of state (10^{-79} Jm^6)
Ne	0.39	0	0	0	4	4	4
CH ₄	2.60	0	0	0	102	102	101
HCl	2.63	1.08	11	6	106	123	157
HBr	3.61	0.78	3	4	182	189	207
HI	5.44	0.38	0.2	2	370	372	350
CH ₃ Cl	4.56	1.87	101	32	282	415	509
NH ₃	2.26	1.47	38	10	63	111	162
H ₂ O	1.48	1.85	96	10	33	139	174

from:

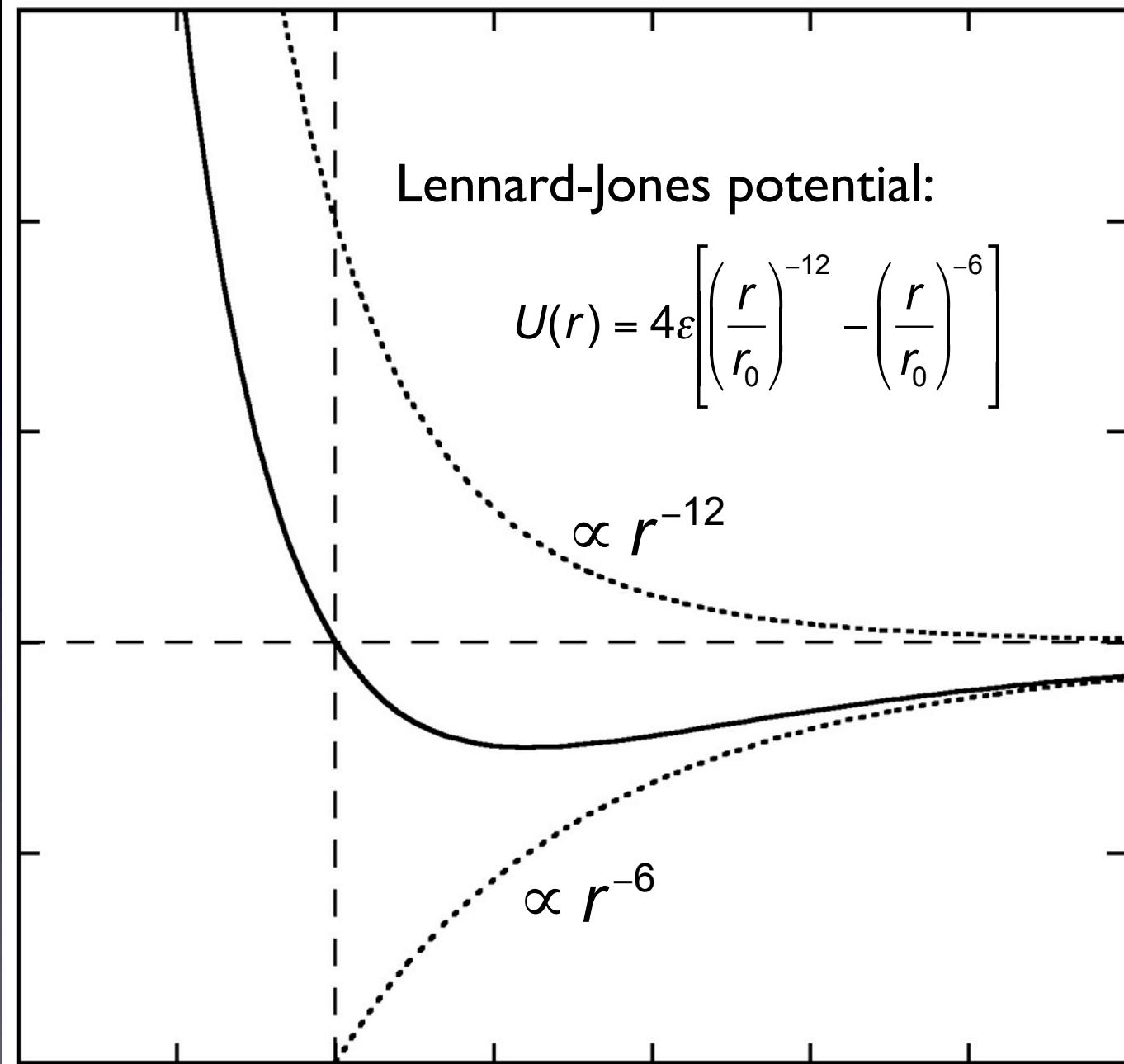
Israelachvili, Intermolecular & Surface Forces

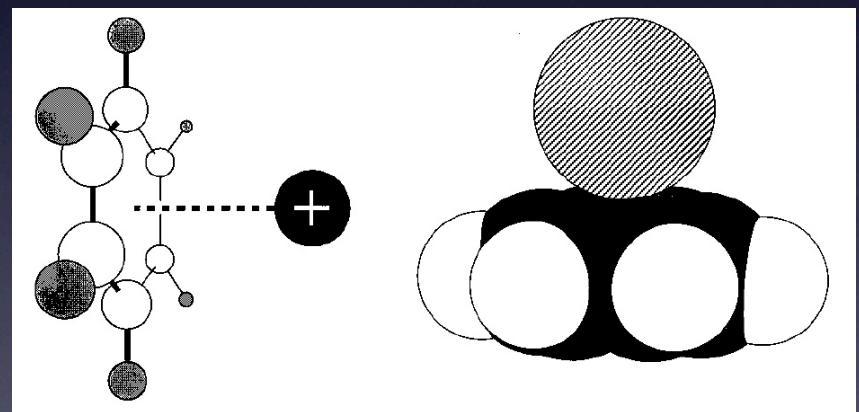
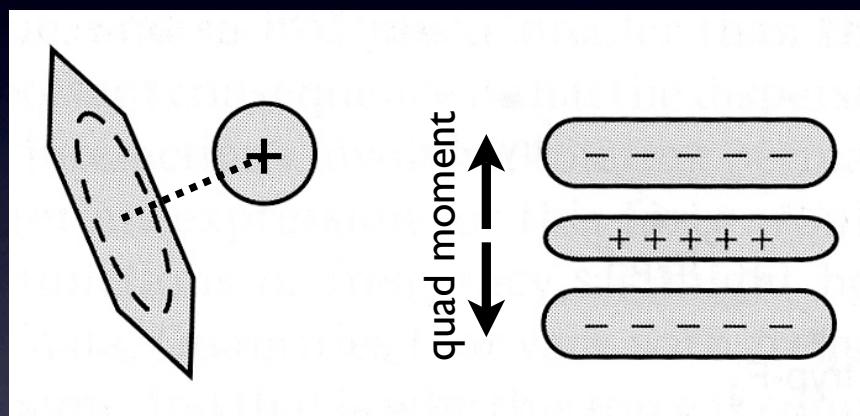
Lennard-Jones potential:

$$U(r) = 4\epsilon \left[\left(\frac{r}{r_0} \right)^{-12} - \left(\frac{r}{r_0} \right)^{-6} \right]$$

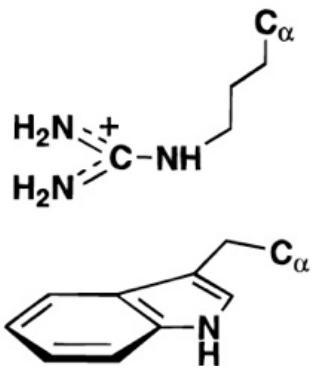
$\propto r^{-12}$

$\propto r^{-6}$

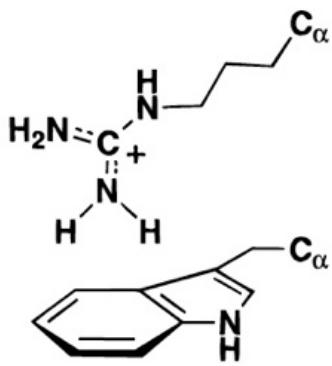




**Parallel
(stacking)**



**Perpendicular
(Hydrogen bonding)**



Arginine (Arg)

Tryptophan (Trp)

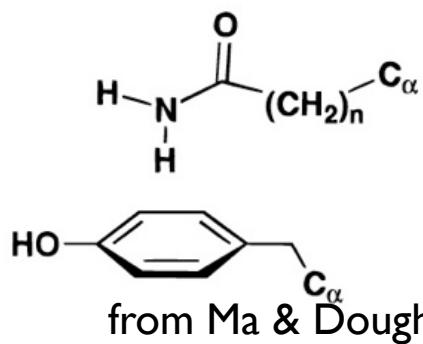
Histidine (His)

Phenylalanine (Phe)

$n = 1$: Asparagine (Asn)

$n = 2$: Glutamine (Gln)

Tyrosine (Tyr)



from Ma & Dougherty, *Chem. Rev.* 97 (1997)